Safety Data Sheet

**Maleic Anhydride (briquettes)** 

# RCT

## **Rierden Chemical & Trading Company**

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## **IDENTIFICATION**

Synonyms	briquettes of: 2,5-furandione; dihydro-2,5-dioxofuran; maleic acid anhydride; MAA
CAS #	108-31-6
Europe EC #	203-571-6
Material Use	mfg. of unsaturated polyester & alkyd resins, mfg. of petroleum & lubricant additives & etc.

<b>EMERGENCY</b>	INFORMATION	
In the U.S.A.	<b>Call CHEMTREC</b>	(800) 424-9300
In Canada	Call CANUTEC (collect	<i>(613) 996-6666</i>

## HAZARD IDENTIFICATION

GHS Class	skin, eye corrosive	acute oral	acute skin	acute inhal.	resp. sensitizer	skin sensitizer
(Category)	(1B)	(4)	(4)	(4)	(1)	(1)
Signal Words	DANGER	WARNING	WARNING	WARNING	DANGER	WARNING
Hazard Statements	causes severe skin burns & eye damage (H314)	harmful if swallowed (H302)	harmful if in contact with skim (H312)	harmful if inhaled (H332)	may cause allergy or asthma symptoms if inhaled (H334)	may cause an allergic skin reaction (H317)

#### **GHS Precautionary Statements for Labelling**

P260 Do not breathe dust.

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P262 Do not get in eyes, on skin or on clothing.

P264 Wash thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P272 Contaminated work clothing should not be allowed out of the workplace.

P280Wear eye protection, protective gloves and clothing of butylor neoprene.P312Call a poison control centre or doctor if you feel unwell.

- P312Can a poison control centre of doctor if you reeraniwen.P313 & P333If skin irritation or rash occurs, get medical advice/attention.
- P362, P364 Take off contaminated clothing and wash it before reuse.

P304, P340Factor containing and wash it before rease.P304, P340If inhaled remove person to fresh air and keep comfortable for breathing.

P305, P351, P338 Rinse cautiously with water for several minutes. Remove contact lenses if present & easy to do. Continue rinsing.

III	COMPOSITION	CAS NUMBER	%	TLV mg/m <sup>3</sup>	LD <sub>50</sub> (mg/kg) ORAL	LD <sub>50</sub> (mg/kg) SKIN	LC <sub>50</sub> mg/m <sup>3</sup> INHALATION
Maleic Anhydride		108-31-6	>98%	0.01 (skin)	>390	>610	>152



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### IV FIRST AID

- SKIN:Brush off. Then wash with plenty of water. Remove contaminated clothing. Do not reuse until thoroughly<br/>laundered. Seek medical help promptly if there is persistent itching or redness in the affected area.EYES:Wash eyes with plenty of warm water, holding eyelids open. Seek medical assistance if there is any irritation.
- INGESTION: Give plenty of water to dilute product. Do not induce vomiting (*NOTE below*). Keep victim quiet. If vomiting occurs, lower victim's head below hips to prevent inhalation of vomited material. Seek medical help promptly.

**NOTE:** Corrosive substance: first aid must be applied immediately! Inadvertent inhalation of vomited material may seriously damage the lungs. The stomach should only be emptied under medical supervision, after the installation of an airway to protect the lungs.

## FLAMMABILITY & FIRE-FIGHTING

Flash Point Autoignition Temperature	102°C / 215°F (closed cup) 477°C / 890°F
Flammable Limits	1.4% - 7.1%
Combustion Products Firefighting Precautions	carbon monoxide, nitrogen oxides, smoke, acetylene, part oxidised hydrocarbon fragments foam, CO <sub>2</sub> , water fog or spray; firefighters must wear SCBA
Static Discharge	product dust can accumulate a static charge, static discharge could cause ignition of a dust cloud

## VI ACCIDENTAL RELEASE MEASURES

 Leak Precaution
 not applicable – solid material

 Handling Spill
 sweep, shovel & store in closed containers for disposal

 NOTE: wear an efficient respirator with dust filter when dealing with spill!

## VII HANDLING & STORAGE

To limit risk, store the minimum acceptable amount on premises. Keep in clearly identified outdoor storage, preferably separated from the workplace. Maleic anhydride is deliquescent & reacts with warm water to form maleic acid. This reaction can be vigorous! Keep moisture out of storage containers. Avoid alkalis, sources of ignition & oxidising agents (*see Part X*).

This product should be used in enclosed reaction equipment. No liquid, dust or vapour should be allowed to escape the reaction chamber or the storage vessels.

Avoid creating or breathing product dust when handling. Cut, instead of tearing bags. If dust might form in the workplace, install ventilation to control the airborne concentration to regulated limits (*Part VIII, below*). If dealing with a spill, wear a suitable respirator with an efficient dust filter. Avoid all contact with skin & wash work clothes frequently. An eye bath & safety shower must be available near the workplace.

## VIII EXPOSURE CONTROL & PERSONAL PROTECTION

ACGIH TLV	0.01mg/m <sup>3</sup> (skin & respiratory sensitizer)	ACGIH STEL	not listed
OSHA PEL	$1.0 \text{mg/m}^3$	OSHA STEL	not listed
Ventilation	always use in closed equipment; mechanica	l ventilation may b	be required to control airborne mist concentration
	to regulated limits; all workers should have	a respirator with	dust filter for escape should containment or
	ventilation fail; if handling occurs at elevate	ed temperature, us	e a dust filter & an organic vapour cartridge
Hands	wear butyl or neoprene gloves - other types	also protect; conf	firm suitability with supplier
Eyes	safety glasses with side shields or chemical	goggles - always	protecteyes!
Clothing	special protective clothing is not normally	required because	contact with the product should never occur;
	if contact is possible - eg: when equipment	is charged with p	roduct – wear appropriate chemically resistant
	(hands, above) protective garments such as	apron, boots, long	sleeves, face shield, etc

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## PHYSICAL AND CHEMICAL PROPERTIES

NOTE: for Flash Point, Autoignition Temp, & Flammable Limits see Part 5.

Odour & Appearance	white crystalline solid (briquettes) with irritating, acrid, choking odour; deliquescent,
	absorbing moisture from the air forming a wet surface on the briquette
Odour Threshold	1ppm – 1.3ppm – odour gives unreliable warning to presence of toxic vapour

Vapour Density (air $= 1$ )	3.4
Decomposition Temperature	370°C / 698°F
Boiling Point	202°C / 396°F
Melting Point	53°C / 127°F – sublimes readily at this temperature
Specific Gravity	1.48 (20/20°C)
Water Solubility	163 grams per litre (30°C / 86°F) - reacts with water (slowlywith cold) to form maleic acid
- in other solvents	acetone, ethyl acetate, benzene, chloroform; poor solubility in aliphatic hydrocarbons
Log Po/w (Octanol/H2O Partition Coefficient)	not available – maleic anhydride reacts with water; Maleic Acid $LogP_{o/w} = -0.55$
Viscosity	not applicable – <i>solid material</i>
pH (maleic acid solution)	2.4 (0.01 molar solution) – acid pH due to formation of maleic acid insolution
Conversion Factor	$1 \text{ppm} - 4.0 \text{mg/m}^3$
Molecular Weight	98 grams/mole

## REACTIVITY

Dangerously Reactive With

Also Reactive With

Chemical Stability

Mechanical Impact

IX

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strong oxidising agents, light metals (Na, Ca, K, Ba . . .), alkalis, amines, strong reducing agents, ammonia, ammonia solutions and salts; hot water (rapid, heat-producing reaction) cold water, alcohols to form esters; sodium or potassium carbonates; solutions (maleic acid) corrosive to steel (but not stainless 304, 316), some aluminum alloys & zinc NOTE: Maleic anhydride is a reactive substance. Check all other substances in the workplace for their compatibility with maleic anhydride. normally stable; co-polymerises with ethylene, propylene, diethylene Decomposes in Presence of moisture: slowly when cold, rapidly (with heat evolution) when hot **Decomposition Products** maleic acid not sensitive

#### XI TOXICITY INFORMATION

#### i. ACUTE EXPOSURE

Skin Contact	irritating if contact persists, particularly in the presence of moisture; <i>maleic anhydride is deliquescent</i> , <i>readily picking up moisture from air to become (corrosive, acidic) maleic acid</i> ; corrosive <sup>1</sup>
Skin Absorption	yes, slowly; toxic effects unlikely by this route
Eye Contact	severely & rapidly irritating (strongly acidic) on contact with moist eye surface, blindness is possible; vapour irritating at 1.0-1.5mg/m <sup>3</sup> ; corrosive <sup>1</sup>
Inhalation	vapour or dust severely irritating to respiratory system; coughing, shortness of breath & possible pulmonary oedema ( <i>dusting limited by briquette format; also, deliquescence limits dust formation.</i> ) respiratory sensitizer <sup>1</sup>
Ingestion	contact with any part of the digestive system (mouth, throat) likely to produce corrosive burns
LD <sub>50</sub> (oral)	400, 625, 824-900, 1030 <sup>1</sup> , 1050 & 1090 <sup>1</sup> mg/kg (rat), 465mg/kg (mouse), 875mg/kg (rabbit), 390mg/kg (guinea pig)
LD50 (skin) LC50 (inhalation)	$610 \text{mg/kg} (\text{rat}), > 631 \& 2620 \text{mg/kg} (\text{rabbit})^1, > 20,000 \text{mg/kg} (\text{guinea pig}) - severe burns, no mortality 152 \text{mg/m}^3 (\text{rat}), 4350 \text{mg/m}^3 (\text{rat})^1$

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## TOXICITY INFORMATION, cont'd

#### ii. CHRONIC EXPOSURE

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Generalprolonged or repeated *exposure to vapour* may cause skin redness and dermatitis; pain & redness of<br/>eyes also associated with chronic exposure to vapour; chronic damage to cornea also reportedSensitisingrespiratory sensitisation may follow exposure to vapour; skin sensitisation to dust may also occur<br/>not known to be a tumorigen or a carcinogen in humans or animals1<br/>no known effect on humans or animals1

## XII ECOLOGICAL INFORMATION

Bioaccumulation Biodegradation	cannot bioaccumulate due to rapid (0.5 sec) hydrolysis & biodegradation <i>as maleicacid</i> <i>as maleic acid</i> , maleic acid biodegrades rapidly in the presence of oxygen; above 40% in 5 days; one report claims 98% in several hours; also 90% in 25 days, 93% in 21 days, 93% in 11 days & others <sup>1</sup>
Abiotic Degradation	reacts with atmospheric hydroxyl (OH) radicals; <sup>1</sup> / <sub>2</sub> -life in air from 5-16hr to 5-7 days, also 8 hours <sup>1</sup>
Mobility in soil, water	water soluble; moves readily through soil & the water column
Aquatic Toxicity	for maleic acid (maleic anhydride rapidly reacts with water to become maleic acid)
LC50 (Fish, 96 hr)	5mg/litre (Pimephelas promelas); 75mg/litre (Oncorhynchus mykiss & Lepomis macrochirus) <sup>1</sup>
LC <sub>50</sub> (Crustacea, 48hr)	$43^1$ , $83 \& 160 \text{ mg/l}$ (Daphnia magna); $5600 \text{ mg/litre}$ (Daphnia magna $-pH=7$ )
EC <sub>10</sub> (Algæ)	125mg/litre (Hematococcus pluvialis), 74 & 150mg/litre (Pseudokirchnerella subcapitata) <sup>1</sup>
EC10 (Bacteria)	11,800 & 14,600mg/litre (Pseudomonas putida) <sup>1</sup>

**NOTE:** Maleic anhydride reacts with water to form maleic acid which biodegrades rapidly. Much of the aquatic toxicity ascribed to maleic anhydride may be caused by the low pH due to maleic acid formation.

## XIII DISPOSAL CONSIDERATIONS

Waste Disposal **do not flush to sewer**; may be incinerated in approved facility – rotary kiln (cement facility) above 800°C, or fluidised bed furnace above 450°C; <u>should be handled by a licensed hazardous waste disposal specialist</u>

Containers Drums should be reused. Recondition and pressure test by a licensed reconditioner prior to re-use. Pails must be vented and thoroughly dried prior to crushing and recycling.

IBCs (intermediate bulk containers): polyethylene bottle must be pressure tested & recertified at 30 months.
 Replace at 60 months (5 years). Steel containers must be inspected, pressure tested & recertified every 5 years.
 Warning: never cut, drill, weld or grind on or near this container, even if empty.

## XIV TRANSPORT INFORMATION

USA 49 CFR & Canada TDG Product Identification Number Shipping Name Classification Marine Pollution

Reportable Quantity (RQ)

UN – 2215 maleic anhydride Class 8; Packing Group III *not a marine pollutant* 

5000lbs (2270kg)



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#### REGULATIONS

Canada DSL	on inventory
U.S.A. TSCA	ACTIVE
Europe EINECS	on inventory & REACH registered
Japan ENCS	on inventory #1053 – low molecular weight chain-like organic compounds
Australia AICS	on inventory from June 1996
Korea ECL	on inventory from January 1997
Philippines PICCS	on inventory
New Zealand NZloC	on inventory – <i>approval #HSR 003012, June 2006</i>
Asia-PAC	on inventory

#### U.S.A. Regulations:

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Immediately Dangerous to Life or Health: 10 mg/cu m

Allowable Tolerances: Maleic anhydride is exempted from the requirement of a tolerance when used as a for pesticide formulations applied to apples with a minimum pre-harvest interval of 21 days, in accordance with good agricultural practice as inert (or occasionally active) ingredients in pesticide formulations applied to growing crops only.

OSHA Standards: Permissible Exposure Limit: Table Z-1 8-hr Time Weighted Avg: 0.25 ppm (1 mg/cu m).

NIOSH Recommendations: Recommended Exposure Limit: 10 Hr Time-Weighted Avg: 1 mg/cu m (0.25 ppm).

Threshold Limit Values: 8 hr Time Weighted Avg (TWA): 0.1 ppm, sensitizer. Excursion Limit Recommendation: Excursions in worker exposure levels may exceed 3 times the TLV-TWA for no more than a total of 30 minutes during a work day, and under no circumstances should they exceed 5 times the TLV-TWA, provided that the TLV-TWA is not exceeded. A4; Not classifiable as a human carcinogen.

Atmospheric Standards: This action promulgates standards of performance for equipment leaks of Volatile Organic Compounds (VOC) in the Synthetic Organic Chemical Manufacturing Industry (SOCMI). The intended effect of these standards is to require all newly constructed, modified, and reconstructed SOCMI process units to use the best demonstrated system of continuous emission reduction for equipment leaks of VOC, considering costs, non air quality health and environmental impact and energy requirements. Maleic anhydride is produced, as an intermediate or a final product, by process units covered under this subpart. Listed as a hazardous air pollutant (HAP) generally known or suspected to cause serious health problems. The Clean Air Act, as amended in 1990, directs EPA to set standards requiring major sources to sharply reduce routine emissions of toxic pollutants. EPA is required to establish and phase in specific performance based standards for all air emission sources that emit one or more of the listed pollutants. Maleic anhydride is included on this list.

Clean Water Act Requirements: Maleic anhydride is designated as a hazardous substance under section 311(b)(2)(A) of the Federal Water Pollution Control Act and further regulated by the Clean Water Act Amendments of 1977 and 1978. These regulations apply to discharges of this substance. This designation includes any isomers and hydrates, as well as any solutions and mixtures containing this substance.

CERCLA Reportable Quantities: Persons in charge of vessels or facilities are required to notify the National Response Center (NRC) immediately, when there is a release of this designated hazardous substance, in an amount equal to or greater than its reportable quantity of 5000 lb or 2270 kg. The toll free number of the NRC is (800) 424-8802. The rule for determining when notification is required is stated in 40 CFR 302.4 (section IV. D.3.b).

TSCA Requirements: Pursuant to section 8(d) of TSCA, EPA promulgated a model Health and Safety Data Reporting Rule. The section 8(d) model rule requires manufacturers, importers, and processors of listed chemical substances and mixtures to submit to EPA copies and lists of unpublished health and safety studies. Maleic anhydride is included on this list.

RCRA Requirements: As stipulated in 40 CFR 261.33, when maleic anhydride, as a commercial chemical product or manufacturing chemical intermediate or an off-specification commercial chemical product or a manufacturing chemical intermediate, becomes a waste, it must be managed according to Federal and/or State hazardous waste regulations. Also defined as a hazardous waste is any residue, contaminated soil, water, or other debris resulting from the cleanup of a spill, into water or on dry land, of this waste. Generators of small quantities of this waste may qualify for partial exclusion from hazardous waste regulations (40 CFR 261.5).

#### SARA

#### Physical Hazards

#### Chemical Hazards

□Explosive	⊠Acute toxicity (any route of exposure)
□Flammable	Skin corrosion or irritation
□Oxidizer (liquid, solid or gas)	⊠Serious eye damage or eye irritation
□Self-reactive	☑Respiratory or skin sensitization
□Pyrophoric (liquid or solid)	□Germ cell mutagenicity
□Pyrophoric Gas	Carcinogenicity
□Self-heating	□Reproductive toxicity
□Organic peroxide	Specific target organ toxicity (single or repeated ex.)
Corrosive to metal	□Aspiration hazard
□Gas under pressure (compressed gas)	□Simple Asphyxiant
□In contact with water emits flammable gas	□Hazard Not Otherwise Classified
□Combustible Dust	
Hazard Not Otherwise Not Otherwise Classified	

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EMERGENCY INFORMATION: Call CH

Call CHEMTREC (800) 424-9300

## XVI OTHER INFORMATION

**Date of Preparation Date of Revision** 

#### June 2011 May 2013, March 2016, February 2019, February 2022

Prepared for Rierden Chemical & Trading Company, by Peter Bursztyn

With data from Registry of Toxic Effects of Chemical Substances (RTECS - USA), Hazardous Substance Data Base (HSDB - USA), Cheminfo (CCOHS - Canada), OSHA website, European Chemicals Agency (EChA) dossiers & other sources (below if used), as required/available.

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